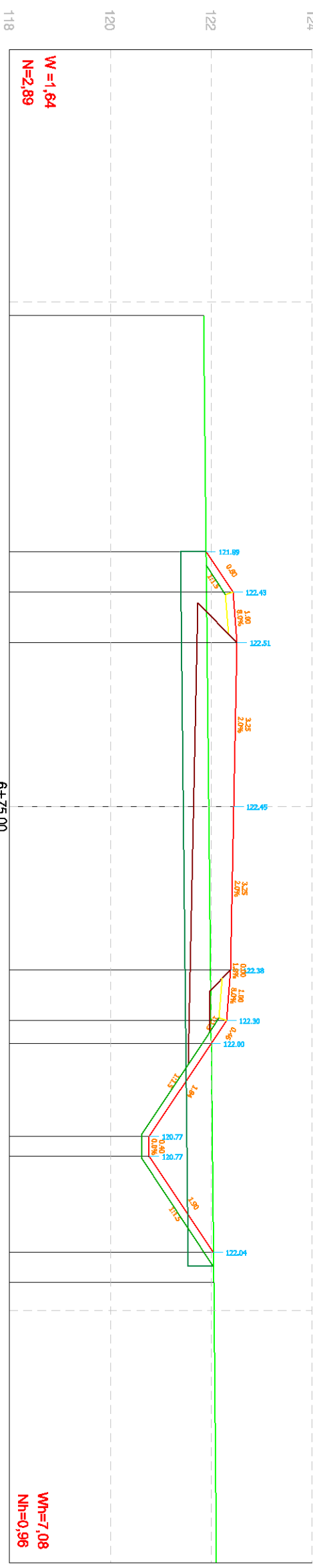
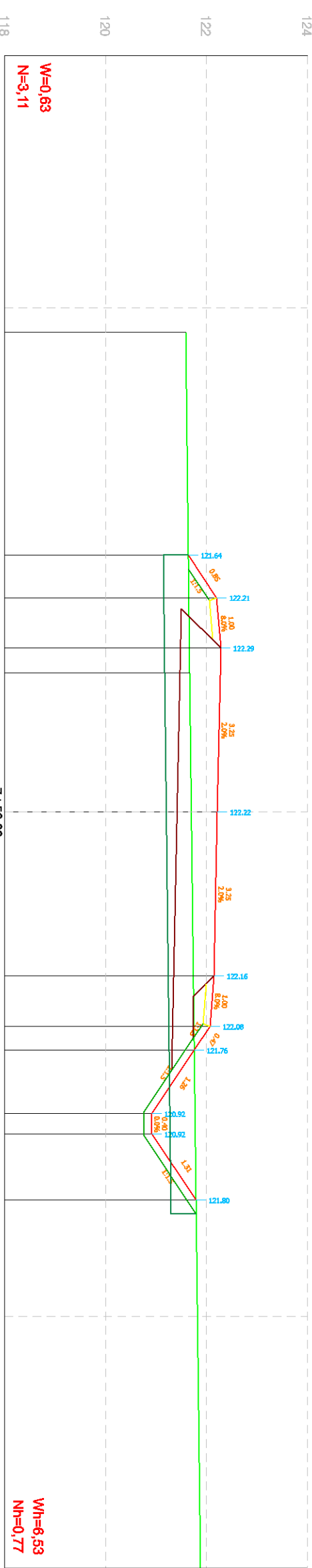
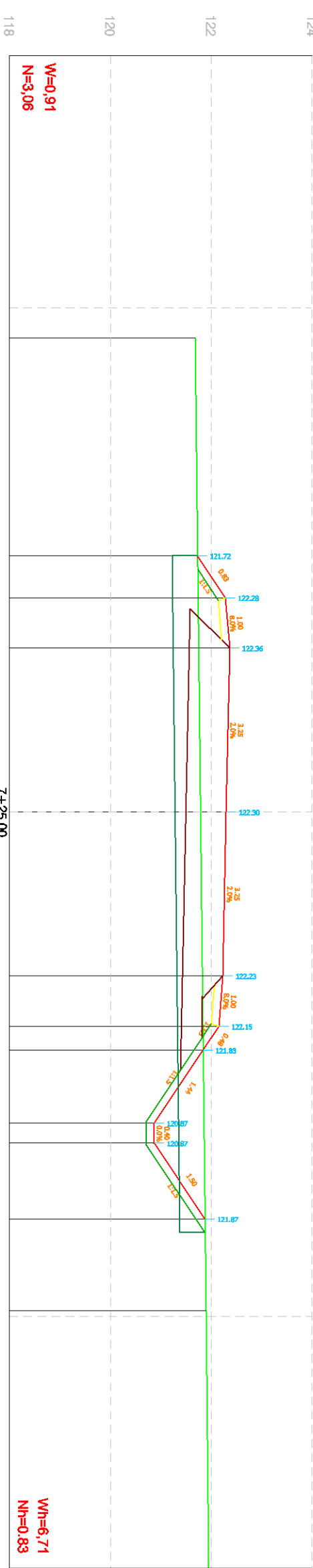
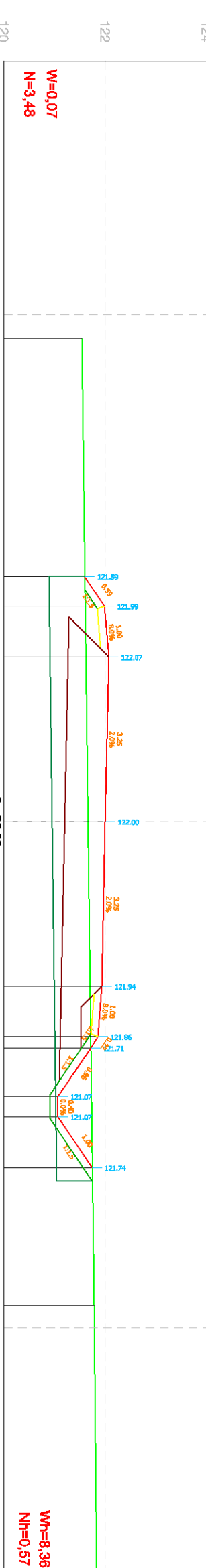
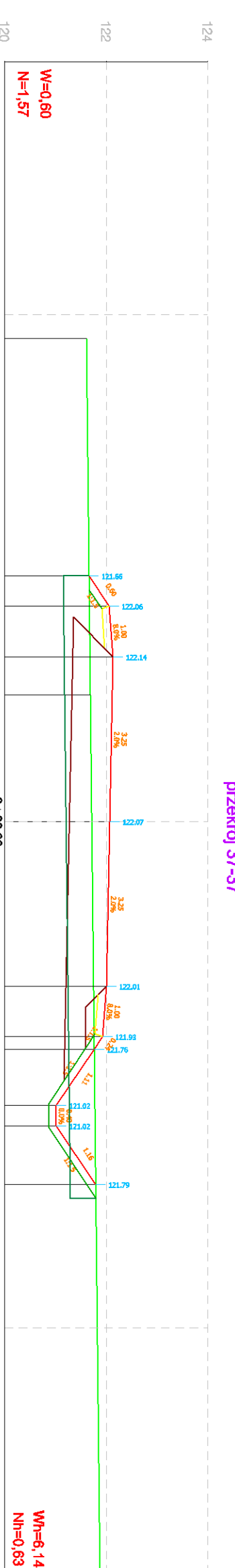
[illegible][illegible]

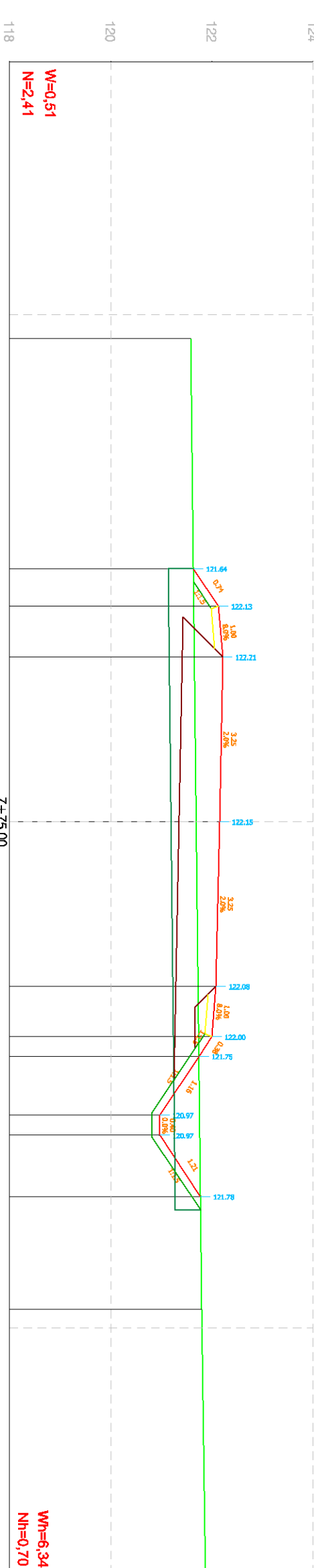
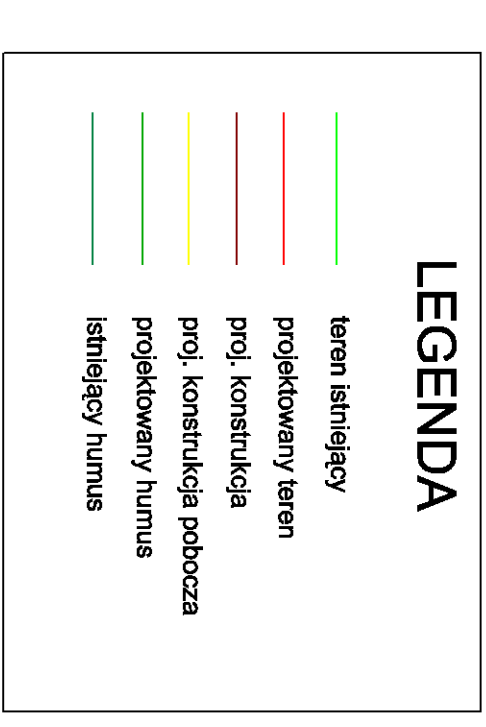
Reaction	Reaction Energy (kJ/mol)	Reaction Entropy (J/mol·K)	Reaction Enthalpy (kJ/mol)
Reaction 1: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$	-285.8	-163.2	-285.8
Reaction 2: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$	-136.0	-102.4	-136.0
Reaction 3: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}$	241.8	163.2	241.8
Reaction 4: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$	0.0	0.0	0.0
Reaction 5: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_3$	141.8	102.4	141.8
Reaction 6: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_4$	281.8	163.2	281.8
Reaction 7: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_5$	421.8	102.4	421.8
Reaction 8: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_6$	561.8	163.2	561.8
Reaction 9: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_7$	701.8	102.4	701.8
Reaction 10: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_8$	841.8	163.2	841.8
Reaction 11: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_9$	981.8	102.4	981.8
Reaction 12: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{10}$	1121.8	163.2	1121.8
Reaction 13: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{11}$	1261.8	102.4	1261.8
Reaction 14: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{12}$	1401.8	163.2	1401.8
Reaction 15: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{13}$	1541.8	102.4	1541.8
Reaction 16: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{14}$	1681.8	163.2	1681.8
Reaction 17: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{15}$	1821.8	102.4	1821.8
Reaction 18: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{16}$	1961.8	163.2	1961.8
Reaction 19: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{17}$	2101.8	102.4	2101.8
Reaction 20: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{18}$	2241.8	163.2	2241.8
Reaction 21: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{19}$	2381.8	102.4	2381.8
Reaction 22: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{20}$	2521.8	163.2	2521.8
Reaction 23: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{21}$	2661.8	102.4	2661.8
Reaction 24: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{22}$	2801.8	163.2	2801.8
Reaction 25: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{23}$	2941.8	102.4	2941.8
Reaction 26: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{24}$	3081.8	163.2	3081.8
Reaction 27: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{25}$	3221.8	102.4	3221.8
Reaction 28: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{26}$	3361.8	163.2	3361.8
Reaction 29: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{27}$	3501.8	102.4	3501.8
Reaction 30: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{28}$	3641.8	163.2	3641.8
Reaction 31: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{29}$	3781.8	102.4	3781.8
Reaction 32: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{30}$	3921.8	163.2	3921.8
Reaction 33: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{31}$	4061.8	102.4	4061.8
Reaction 34: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{32}$	4201.8	163.2	4201.8
Reaction 35: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{33}$	4341.8	102.4	4341.8
Reaction 36: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{34}$	4481.8	163.2	4481.8
Reaction 37: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{35}$	4621.8	102.4	4621.8
Reaction 38: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{36}$	4761.8	163.2	4761.8
Reaction 39: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{37}$	4901.8	102.4	4901.8
Reaction 40: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{38}$	5041.8	163.2	5041.8
Reaction 41: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{39}$	5181.8	102.4	5181.8
Reaction 42: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{40}$	5321.8	163.2	5321.8
Reaction 43: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{41}$	5461.8	102.4	5461.8
Reaction 44: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{42}$	5601.8	163.2	5601.8
Reaction 45: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_{43}$	5741.8	102.4	5741.8



Reine p. d. Salzwasser	Oxidation:	Reine Luft/Luft	Oxidation:
	5.08	121.68	
	4.25		
	3.25		
	0.00		
	3.25		
	4.25		
	4.73		
	6.07		
	6.57		
	8.07		
		121.69	
		9.88	
		15.00	

[illegible]

<i>Predictor performance:</i>	<i>Obligations:</i>	<i>Predictor satisfaction</i>	<i>Obligations:</i>
-0.85	12166	-4.85	12166
-4.25	12206	-4.25	12206
-3.25	12204	-3.25	12204
-2.50	12169	-2.50	12169
0.00	12207	0.00	12207
3.25	12208	3.25	12208
4.25	12187	4.25	12187
5.61	12102	5.61	12102
6.01	12102	6.01	12102
7.86	12179	7.86	12179
15.00	12188	15.00	12188

[illegible]

HAL-SAN		UWAGA
ZAKŁAD PROJEKTOWY	HALSKI ZBIENIEW	ZP
OBIEKT	PRZEBUDOWA DROGI-UL. WISNIOWEJ NA ODCINKU SMOLEC - PRZEMOŁ GMINA KĄTY WROCŁAWSKIE	EA
PRZYSŁOWIEK	PRZEMOŁ CHARAKTERYSTYCZNE	MA
STADIUM	PROJEKT WYKONAWCZY	S
WZNAJOSHO	WZNAJOSHO SŁOŻENIEM 2,000/01	DATA
PROJEKTOWY	PROJEKTOWY	10.02.02
PROJEKTOWY	PROJEKTOWY	S